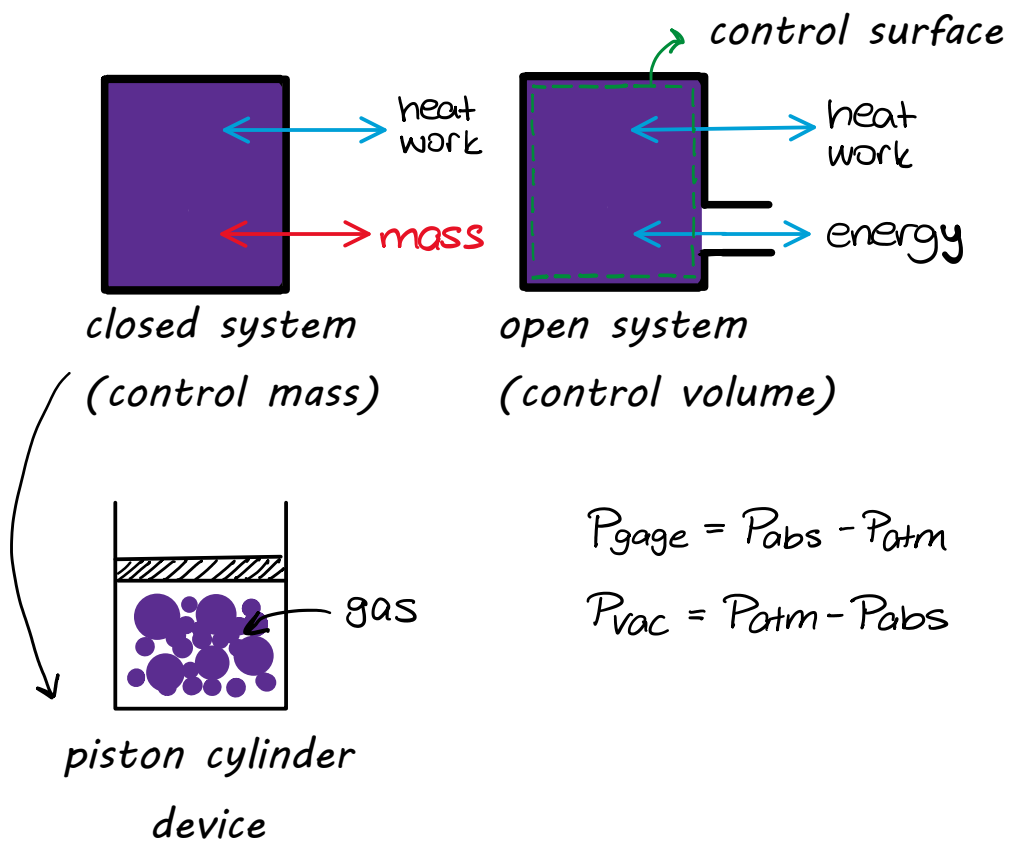


basic concepts

11 Şubat 2025 Salı

12:16



$$P_{\text{gage}} = P_{\text{abs}} - P_{\text{atm}}$$

$$P_{\text{vac}} = P_{\text{atm}} - P_{\text{abs}}$$

P_{atm} = outdoor air pressure
↳ barometric "

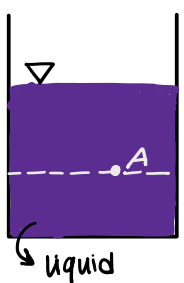
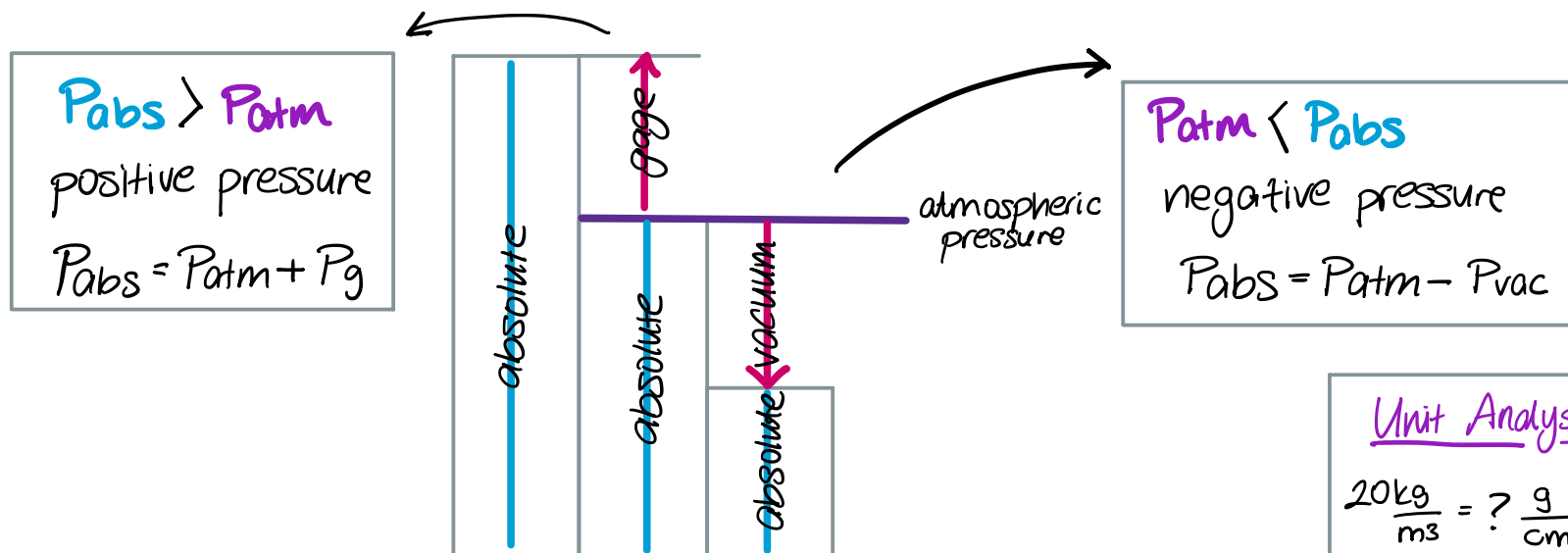
P_{gage} = effective pressure
↳ manometric "

there are two types of properties

- intensive: independent of mass (s, P, T)
- extensive: dependent of mass (E, V)

$$\hookrightarrow \frac{E}{m} = e, \frac{V}{m} = v, \frac{1}{s} = v$$

two intensive properties \rightarrow known state



$$P_A = \rho_{\text{liq}} \cdot g \cdot h$$

$$P_{\text{abs}} = P_{\text{atm}} + P_g$$

$$(P_{\text{atm}} = 101.3 \text{ kPa} = 760 \text{ mmHg})$$

Unit Analysis:

$$20 \frac{\text{kg}}{\text{m}^3} = ? \frac{\text{g}}{\text{cm}^3} \quad 20 \frac{10^3 \text{g}}{10^6 \text{cm}^3} = 2 \times 10^{-2} \frac{\text{g}}{\text{cm}^3}$$

$$5 \text{ bar} = ? \frac{\text{N}}{\text{m}^2} \quad (1 \text{ bar} = 100 \text{ kPa}) \quad 5 \text{ bar} = 5 \times 10^5 \text{ Pa} = \frac{\text{N}}{\text{m}^2}$$

$$5 \text{ cl} = ? \text{ liters} \rightarrow 5 \times 10^{-2} \text{ L}$$

$$100 \text{ km/h} = ? \text{ m/s} \quad \frac{100 \times 10^3 \text{ m}}{3600 \text{ sec}} = \frac{10^3}{36} \approx 27.7 \text{ m/s}$$

low altitude
high pressure
boiling point \uparrow
freezing point \downarrow

high altitude
low pressure
boiling point \downarrow
freezing point \uparrow

